

## Section 11 1 Describing Chemical Reactions Pages 321 329

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SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

### SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS 1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water. 2.

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A chemical reaction can be concisely represented by a chemical 1.. The substances that undergo a chemical change are the 2.. The new substances formed in a chemical reaction are the 3.. In accordance with the law of conservation of , 4. a chemical equation must be balanced. When balancing an 5. equation, you place in front of reactants and products so 6.

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Section 11.1 Describing Chemical Reactions Worksheet Answers - When you find a template that you would like to use, begin customizing it immediately and you may also double-click the template thumbnail to open it! You will discover a number of the templates are completely free to use and others call for a premium account.

### Section 11.1 Describing Chemical Reactions Worksheet Answers

One of the best worksheets for Section 11.1 Describing Chemical Reactions is the chapter from "It's A Molecule" by Stephen E. Gries, D.Sc. This book contains a review of the molecules found in everyday things. In addition, the book tells about the different types of molecules and explains why some have an effect on each other.

### Section 11.1 Describing Chemical Reactions Worksheet Answers

322 Chapter 11 Figure 11.2 Three common chemical reactions are shown. When methane gas burns, it combines with oxygen to form carbon dioxide and water. Iron turns to red-brown rust (iron(III) oxide) in the presence of oxygen in the air. Hydrogen peroxide decomposes to water and oxygen when used as an antiseptic.

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Section 11.1 - Describing Chemical Reactions In a chemical reaction, the reactants are written on the left and the products on the right. The arrow that separates them is called yield. Reactants  $\rightarrow$  Products

#### Chapter 11: Chemical Reactions

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Patterns of chemical behavior will become apparent and allow you to predict the products of reactions. Combination ReactionsThe first type of reaction is the combination, or synthesis, reaction. A combination reaction is a chemical change in which two or more substances react to form a single new substance.

#### 11.2 Types of Chemical Reactions - Evaluation 2016

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