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Using the methods developed in class:  
Step 1: Multiply both sides by the

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partition function  $H = (E + pV) e^{E_j p V_j} V_j$ .  
Step 2: Get the temperature derivative  
at constant  $(N, P)$  (The conjugate  
variable to  $H$  in this case)  $\frac{\partial H}{\partial T} = 1 + H$   
 $(E + pV) e^{E_j p V_j} = 1 (E_j + pV) e^{E_j p V_j} \cdot \frac{\partial T}{kT^2}$   
 $kT^2 kT^2 N, P V_j V_j$ .

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atom. Show that in the Thompson hydrogen atom, the net force exerted on the electron is directly proportional to its displacement from the center of the atom (i.e., it undergoes simple harmonic motion).

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following:  $\Omega = \frac{1}{h^3} \int \exp(-\beta H) d^3p d^3q$  and  $q = \int \frac{1}{h^3} \exp(-\beta H) d^3p d^3q$   
 $h^3$  is the partition function for the grand canonical ensemble, where  $T, V$ , are fixed.

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