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CHAPTER 1 REVIEW Matter and Change SECTION 3 SHORT ANSWER Answer the following questions in the space provided.

1. A horizontal row of elements in the periodic table is called a(n) period .
2. The symbol for the element in Period 2, Group 13, is B .
3. Elements that are good conductors of heat and electricity are metals .
- 4.

1 Matter and Change - HUBBARD'S CHEMISTRY

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

7 Chemical Formulas and Chemical Compounds

CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT

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ANSWER Answer the following questions in the space provided.

1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

6 Chemical Bonding

CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER

Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral atom is the atom's

5 The Periodic Law

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CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2.

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6.0 mol of N₂ are mixed with 12.0 mol of H

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions
Teacher Notes and Answers Chapter 12 SECTION 1 SHORT
ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The
mixture is a colloid. The properties are consistent with those
reported in Table 3 on page 404 of the text. The particle size is
small, but not too small, and the mixture

CHAPTER 12 REVIEW Solutions

Modern Chemistry Chapter 7 Section 1 Review. STUDY.
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Created by. dolphinking. study the thing. Terms in this set (25)
Chemical formula. indicates the relative number of atoms of
each kind in a chemical compound. Reveals the number of
atoms of each element contained in a single molecule ...

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter
SECTION 3 SHORT ANSWER Answer the following questions in
the space provided. 1. Explain the difference between the mass
number and the atomic number of a nuclide. Mass number is the
total number of protons and neutrons in the nucleus of an
isotope.

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