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SECTION 6.1 Nature favors arrangements in which potential energy is minimized. For example, a boulder is less likely to balance at the top of a hill than it is to roll to the bottom of a valley. The boulder at the top of the hill is not stable. Atoms are usually not stable when isolated. They usually combine to form more stable arrangements of matter.

## **CHAPTER 6 Chemical Bonding**

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Modern Chemistry 1 Chemical Bonding  
CHAPTER 6 Chemical Bonding SECTION  
1 Introduction to Chemical Bonding  
OBJECTIVES 1. Define Chemical bond. 2.  
Explain why most atoms form chemical  
bonds. 3. Describe ionic and covalent  
bonding.. 4.

**CHAPTER 6 Chemical Bonding**

CHAPTER 6 REVIEW Chemical Bonding  
SECTION 3 SHORT ANSWER Answer the  
following questions in the space  
provided. 1. a The notation for sodium  
chloride, NaCl, stands for one (a)  
formula unit. (c) crystal. (b) molecule.  
(d) atom. 2. d In a crystal of an ionic  
compound, each cation is surrounded by  
a number of (a) molecules. (c) dipoles.  
(b) positive ions. (d) negative ions.

**6 Chemical Bonding**

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Modern Chemistry Chapter 6 "Chemical Bonding". chemical bond. covalent bonding. ionic bonding. nonpolar-covalent bond. a mutual electrical attraction between the nuclei and valence.... a chemical bond resulting from the sharing of an electron pair.... the chemical bond resulting from electrical attraction between....

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

**3 Atoms: The Building Blocks of Matter**

CH 1 Reading Assignment Modern Chemistry. CH 1 Vocabulary-New. CH 1 Mixed Questions. CH 1 Matter & Energy Vocabulary-New ... Notes: Chapter 6 CH 6 Chemical Bonding . Reference: CH 6 Common Ion Table. ... Ch 6 Section Review 6.1 & 6.2. Ch 6 Section Review 6.3. Ch 6 Section Review 6.4.

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