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Ionic bonding is when two ions of opposite charge attract and bond to each other forming an ionic compound. Consider the following examples of formulas for ionic compounds:

- One Na^+ (sodium ion) and one Cl^- (chloride ion) bond to make NaCl , "sodium chloride."
- One Mg^{2+} (magnesium ion) and two F^- (fluoride ion) bond to make MgF_2

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Information: Naming Ions - S.W.H.S CHEMISTRY

The means of bonding that we will consider now is called ionic bonding, which occurs between a metal and a nonmetal. As you know, opposite charges attract. Ionic bonding is when two ions of opposite charge attract and bond to each other forming an ionic compound.

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COVALENT BONDING Name Covalent bonding occurs when two or more nonmetals share electrons. attempting to attain a stable octet of electrons at least part of the time. For example: Note that hydrogen is content with 2, not 8. electrons. Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may

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