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I The answer is expressed as 255.5 mL since 46.0 mL has only one decimal

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(Answer Key) Determine the oxidation
number of the elements in each of the
following compounds: a. H_2CO_3 H: +1,
O: -2, C: +4 b. N_2 N: 0 c. $\text{Zn}(\text{OH})_2$ Zn:
2+, H: +1, O: -2 d. NO_2 N: +3, O: -2 e.
 LiH Li: +1, H: -1 f. Fe_3O_4 Fe: +8/3, O:
-2; Identify the species being oxidized
and reduced in each of the ...

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