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10. Possible answers include: white with blue lines, flat, thin, 1-2 g, flammable, solid. 11. Possible answers include: red/green/red & blue, cylindrical solid filled with liquid which has a high concentration of gas, 355 mL, pressure increases if shaken. 12. Possible answers include: color, rectangular. solid, made of paper and cardboard ...

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Glencoe Chemistry - Matter And Change Chapter 18: Acids ...

Section 18.2 Assessment page 649 17.
Describe the contents of dilute aqueous

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solutions of the strong acid HI and the weak acid HCOOH. The solution of HI contains only H^+ and I^- and water molecules. The solution of HCOOH contains H^+ and HCOO^- ions, and HCOOH and H_2O molecules. 18. Relate the strength of a weak acid to the strength of its conjugate base.

Acids and Bases

Section 18.1 • Introduction to Acids and Bases 635
Figure 18.2 The strong acid hydrochloric acid (HCl), also called muriatic acid, is used to clean bricks and concrete. The strong base sodium hydroxide (NaOH) can clear clogged drains. Acids turn blue litmus red. Bases turn red litmus blue.

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