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## Answers Chapter 19

Chapter 2 Matter and Change . Name  
Date Class ... Core Teaching Resources .  
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Practice Problems In your notebook,  
solve the following problems. ... Use the  
Table 2.1 to answer the following  
questions. 6.

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quizzes to this? Prentice hall chemistry book. "CORE TEACHING RESOURCES" also known as -CORE Teaching Resources Prentice Hall Chemistry - Spiral-bound - Teacher's Edition (2005) by Antony C. Wilbraham-

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In your notebook, answer the following

**SECTION 7.1 IONS** For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion.

a. Ba

2. How many valence electrons does each of the following atoms have?

a. gallium b. fluorine c. selenium

3.

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Choose the best answer for each question and write its letter on the line.

11. Another name for homogenous mixture is a. solution. b. matter. 12. The chemical symbol for iron is: c. element. d. mass. 13. Which of the following is a physical property? a. color b. hardness

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48 Core Teaching Resources c. freezing point . all of the above

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Chemistry is the study of the of matter and the 2. that matter undergoes. ...

Answer the following questions in the space provided. 28. Match each activity below to one of the five branches of

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chemistry. ... 14 Core Teaching  
Resources Part C Matching

## **05 CTR ch01 7/9/04 3:24 PM Page 7 CHEMISTRY 1**

protons; neutrons 2. electrons 3.  
neutrons. 4. electrons 5. ionic 6. The two  
main types of chemical bonding are  
ionic and covalent bonding. Ionic bonds



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are formed when a transfer of electrons takes place from one atom to another, and covalent bonds are formed when electrons are shared between atoms.

## **Ch. 2 Answer Key**

experiments; Core Teaching Resource: 45; Small Scale Lab Manual: p.4-5, All 41 experiments C-2.1 Illustrate electron

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configurations by using orbital notation  
for representative elements. SE/TE:  
133-135, 164, 166, 442 TE: Class  
Activity: 134 TR: Core Teaching  
Resource: 117 TECH: Interactive  
Textbook: Simulation 2 C-2.2 Summarize  
atomic properties

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ions Learn with flashcards, games, and more — for free.

## **Chemistry: Chapter 7, section 7.1 Flashcards | Quizlet**

Name Date 11. The number of sublevels in an energy level is equal to the square of the principal quantum number of that energy level. 12. The maximum number

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of electrons that can occupy the fourth principal energy level of an atom is 32.

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carbon has four valence ...

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Chemistry [Antony C. Wilbraham] on

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138 Core Teaching Resources SECTION

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6.2 CLASSIFYING THE ELEMENTS 1. Use the periodic table to write the electron configuration for silicon. Explain your thinking. 2. Use the periodic table to write the electron configuration for iodine.

**Prentice Hall Chemistry Worksheets**  
Terms in this set (...) conversion factor. a

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ratio of equivalent measurements used to convert a quantity from one unit to another. dimension analysis. a technique of problem-solving that uses the units that are part of a measurement to help solve the problem. density. the ratio of the mass of an object to its volume.  
Chemistry Vocab Chapter 2.



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### **3.3 Conversion Problems/ 3.4 Density Flashcards | Quizlet**

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