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Chemistry The Physical Setting 2011

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0 10. 20. 30. 40. 50. 60. 70. 80. 90. 100. Temperature (°C)
Solubility (g solute/100. g H O) 2 150. 140. 130. 120. 110. 100.
90. 80....

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Table I Heats of Reaction at 101.3 kPa and 298 K Table J Activity Series**
Reaction ΔH (kJ)*
 $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{l})$ -890.4
 $\text{C}_3\text{H}_8(\text{g}) + 5\text{O}_2(\text{g}) \rightarrow 3\text{CO}_2(\text{g}) + 4\text{H}_2\text{O}(\text{l})$ -2219.2
 $2\text{C}_8\text{H}_{18}(\text{l}) + 25\text{O}_2(\text{g}) \rightarrow 16\text{CO}_2(\text{g}) + 18\text{H}_2\text{O}(\text{l})$ -10943
 $2\text{C}_2\text{H}_5\text{OH}(\text{l}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{CO}_2(\text{g}) + 4\text{H}_2\text{O}(\text{l})$ -1452

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PHYSICAL SETTING CHEMISTRY Wednesday, June 22, 2011 — 1:15 to 4:15 p.m., only This is a test of your knowledge of chemistry. Use that knowledge to answer all questions in this examination. Some questions may require the use of the Reference Tables for Physical Setting/Chemistry. You are to answer all questions in all parts of

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may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry. 1 Which characteristics will both generally decrease when elements in Period 3 on the Periodic Table are considered in order from left to right? (1) nonmetallic properties and atomic radius (2) metallic properties and atomic radius

Part A Answer all questions in this part.

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may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry. P.S./Chem.-Jan. '19 [2] 1 The results of the gold foil experiment led to the conclusion that an atom is (1) mostly empty space and has a small, negatively charged nucleus (2) mostly empty space and has a small, positively charged nucleus

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Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry. Base your answers to questions 31 and 32 on the information below and on your knowledge of chemistry. A sample of a molecular substance starting as a gas at 206°C and 1 atm is allowed to cool for 16 minutes.

Part A Answer all questions in this part.

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