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NaH_2PO_4 and Na_2HPO_4 in water, in which NaH_2PO_4 produces the acidic ion and Na_2HPO_4 produces the conjugate base ion. (i) Deduce the acid and

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conjugate base ions that make up the phosphate buffer and state the ionic equation that represents the phosphate buffer. [3] (ii) Describe how the phosphate buffer minimizes the effect of the addition of a

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«signal at» 2.2 2.7 «ppm» AND «H
atom/proton of alkyl/CH next to»
aldehyde/CHO present OR «signal at»

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2.2 2.7 «ppm» AND «H atom/proton of»
RCOCH₂- present OR no «signal at» 4.5
6.0 «ppm» AND absence of «H
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This paper was average in terms of difficulty. Some problems may include the stoich question (the volume one) on Topic 1, the shape of that weird molecule on Topic 4, and energetic

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manipulations on Topic 5. But overall, this paper was okay though. Past papers really helped. P.S. Topic 2 question on the number of electrons in NUCLEUS was LIT.

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