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Modern Chemistry 27 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 4. The electron configuration below violates a. the Pauli exclusion principle. b. the Aufbau principle. c. Hund's rule. d. Both (a) and (c) \_\_\_\_ 5. A photon is emitted from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell. b ...

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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Chapter 4 Test Understand the differences in the energies of different orbital sublevels. Know Aufbau principle, Hund's rule, & Pauli exclusion principle. Identify elements using their electron configuration notation. Determine the electron configuration notation of elements.

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### **Chapter 4 Modern Chemistry Arrangement Of Electrons In ...**

Modern Chemistry 4 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_20. In general, which properties are common to elements within a single group on the periodic table? a. atomic number and atomic weight b. chemical properties and ability to bond to other atoms c. color and luster d. boiling point and freezing point \_\_\_\_21.

### **Assessment Chapter Test A**

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