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ATOMIC STRUCTURE  
AND THE PERIODIC  
TABLE CHAPTER 4  
WORKSHEET PART A

Given the following  
isotopes, determine  
the atomic number, the  
mass number, the  
number of protons,

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electrons and  
neutrons. Isotope  
Symbol  $^{131}\text{I}$  S31  
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**|He SAI**

Chapter 4 - atomic structure; Chapter 5 - electrons in atoms (handouts) Chapter 6 - periodic table & trends (handouts) Chapters 7/9 - ionic bonding & naming (handouts) Chapters 8/9 - covalent bonding & chemical names & formulas (handouts) Chapters

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bonding/IMFs

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- moles (handouts)

Chapter 11 - reactions

...

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Section 4.2 Structure of  
the Nuclear Atom One

change to Dalton's

atomic theory is that

atoms are divisible into

subatomic particles:

Electrons, protons, and

neutrons are examples

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of these fundamental particles There are

many other types of particles, but we will

study these three

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Structure - Henry  
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District**

ATOMIC STRUCTURE

AND THE PERIODIC

TABLE CHAPTER 4

WORKSHEET PART A

Given the following

isotopes, determine

the atomic number, the



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mass number, the  
number of protons,  
electrons and  
neutrons. Isotope  
Symbol 131I S31 U  
|He SAI 81W 36Kr SIDU  
37KD Atomic Number  
Mass Number Protons  
Electrons Neutrons  
Isotope Name PARTB.

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matrix 4 while the

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it is all under the skin  
the suspension has  
been reworked with  
lighter arms 160 coil.

**Atomic Structure  
Worksheet Key  
Chapter 4 | Printable**

...

Chapter 4: Atomic  
Structure Worksheet .  
Answer the following  
questions, circle the  
best answer. 1) Which  
of the following are  
isotopes of each other?  
a)  $^{14}\text{C}$  and  $^{14}\text{N}$  b)  $^3\text{H}$

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and  $4\text{He}$  c)  $2\text{H}$  and  $1\text{H}$

d) none of these . 2)

The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is . . .

## **Chapter 4 Atomic Structure**

### **Assessment Answers**

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Structure ...**

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ANSWER SHEET CHAPTER 4

WORKSHEET PART A

Answers Pearson  
Given the following isotopes, determine the atomic number, the mass number, the number of protons, electrons and neutrons

Isotope	Symbol	Atomic Number	Mass Number	Protons	Electrons	Neutrons
I	$^{131}\text{I}$					
U	$^{238}\text{U}$					
He	$^4\text{He}$					
W	$^{184}\text{W}$					
Kr	$^{84}\text{Kr}$					
Si	$^{28}\text{Si}$					
K	$^{39}\text{K}$					
D	$^2\text{D}$					

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WORKSHEET PART A  
Given the following  
isotopes, determine

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the atomic number, the

mass number, the

number of protons,

electrons and neutrons

Isotope Symbol 131i

S31 Us |He SAl 81W

36Kr SIDU 37KD Atomic

Number Mass Number

Protons Electrons

Neutrons Isotope

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**Worksheet Answers**

Chapter 4: Atomic

Structure Worksheet .

Answer the following

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### Atomic Structure

questions, circle the best answer. 1) Which of the following are isotopes of each other?

- a)  $^{14}\text{C}$  and  $^{14}\text{N}$
  - b)  $^3\text{H}$  and  $^4\text{He}$
  - c)  $^2\text{H}$  and  $^1\text{H}$
  - d) none of these
- . 2)

The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is . a)  $5+$  b)  $-3$  c)  $3+$  d)  $-9$

### **Name Chapter 4: Atomic Structure Worksheet Answer the . . .**

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...

This unit's focus is on  
atomic structure.

Students will cover  
concepts related to  
atomic mass, isotopes,  
electron configurations

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and periodicity.  
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**Unit 2 - Chapters 4,  
5 & 6 - Mrs. Gingras'  
Chemistry Page**

Chapter 4 - Atomic  
Structure. Chapter 5 -  
Electrons in Atoms.  
Chapter 6 - The  
Periodic Table. Chapter  
7 - Ionic and Metallic  
Bonding. Chapter 8 -  
Covalent Bonding.  
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Names and Formulas.

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