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a heterogeneous mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension particles) Brownian motion the erratic, random, movements of colloid particles that results from collisions of particles of the dispersion medium with the dispersed particles

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## **Bing**

Chemistry II: Chapter 14 Mixtures and Solutions. a type of heterogeneous mixture whose particles settle out over time and can be separated from the mixture by filtration. a heterogeneous mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension particles).

## **Chemistry II: Chapter 14 Mixtures and Solutions ...**

The Mixtures and Solutions chapter of this Glencoe Chemistry - Matter and Change companion course helps students learn the essential chemistry lessons of mixtures, solvents and solutes.

## **Glencoe Chemistry - Matter And Change Chapter 14: Mixtures ...**

Section 14.1 Collision Theory: A Model for the Reaction Process. Goals. To describe a model, called collision theory, that helps us to visualize the process of many chemical reactions. To use

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collision theory to explain why not all collisions between possible reactants lead to products. To use collision theory to explain why possible reactants must collide with an energy equal to or above a certain amount to have the possibility of reacting and forming products.

### **Chapter 14 - The Process of Chemical Reactions**

Mixtures and Solutions. Section 14.1. Heterogeneous and Homogeneous Mixtures. In your textbook, read about suspensions and colloids. For each statement below, write . true. or . false. \_\_\_\_ 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed. \_\_\_\_ 2. The most abundant substance in a colloid is the dispersion medium.

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tomorrow. The important thing is not to stop questioning.

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